



Volatile Acidity in Wine by Segmented Flow Analysis (SFA)

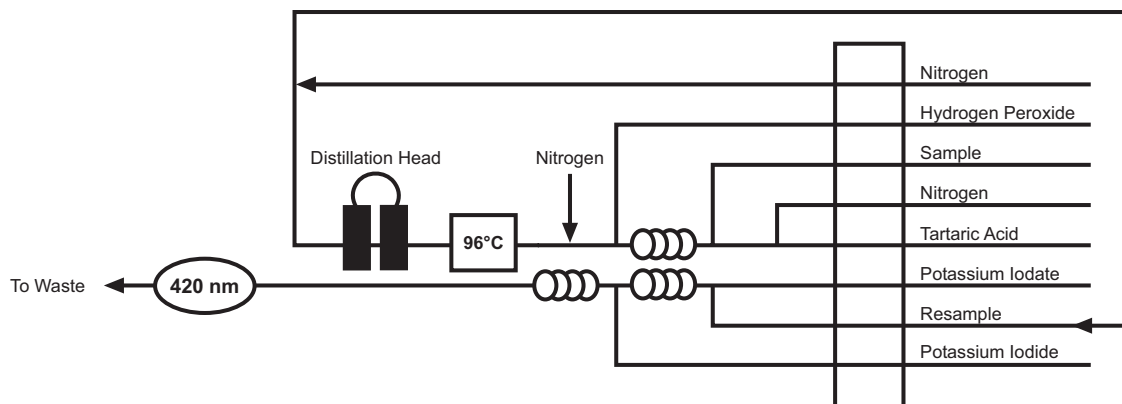
(Cartridge Part #A003133)

1.0 Scope and Application

- 1.1 This method is used for the determination of volatile acidity in red and white wine.
- 1.2 The applicable range of the method is 0.15–1.5 g/L or 0.015–0.15 g/100 mL as acetic acid. The range may be extended to analyze higher concentrations by sample dilution. The Method Detection Limit (MDL) of this method was not determined because volatile acidity concentrations below 0.15 g/L are not of interest in the analysis of wine.

2.0 Summary of Method

- 2.1 The determination of volatile acidity in wine is based upon the reaction of the wine sample with tartaric acid and hydrogen peroxide, combined with distillation at 96°C. Hydrogen peroxide is used to oxidize interfering sulfites (SO_2). The resampled distillate reacts with iodide and iodate, forming a brown-yellow colored complex. The absorbance is measured at 420 nm.
- 2.2 The quality of the analysis is assured through reproducible calibration and testing of the Segmented Flow Analysis (SFA) system.
- 2.3 A general flow diagram of the SFA system is shown below (see Section 17.0 for a detailed flow diagram).



3.0 Definitions

Definitions for terms used in this method are provided in Section 16.0, “Glossary of Definitions and Purposes.”

4.0 Interferences

4.1 Proteins and colored compounds that could interfere are removed by distillation.

5.0 Safety

5.1 The toxicity or carcinogenicity of each compound or reagent used in this method has not been fully established. Each chemical should be treated as a potential health hazard. Exposure to these chemicals should be reduced to the lowest possible level.

5.2 For reference purposes, a file of Material Safety Data Sheets (MSDS) for each chemical used in this method should be available to all personnel involved in this chemical analysis. The preparation of a formal safety plan is also advisable.

5.3 The following chemicals used in this method may be highly toxic or hazardous and should be handled with extreme caution at all times. Consult the appropriate MSDS before handling.

5.3.1 Acetic Acid, glacial, CH_3COOH (FW 60.05)

5.3.2 Ethanol, anhydrous, $\text{C}_2\text{H}_5\text{OH}$ (FW 46.07)

5.3.3 Hydrogen Peroxide, 30% w/v, H_2O_2 (FW 34.02)

5.3.4 Potassium Iodate, KIO_3 (FW 214.00)

5.3.5 Potassium Iodide, KI (FW 166.01)

5.3.6 Tartaric Acid, $\text{C}_4\text{H}_6\text{O}_6$ (FW 150.09)

5.4 Unknown samples may be potentially hazardous and should be handled with extreme caution at all times.

5.5 Proper personal protective equipment (PPE) should be used when handling or working in the presence of chemicals.

5.6 This method does not address all safety issues associated with its use. The laboratory is responsible for maintaining a safe work environment and a current awareness file of OSHA regulations regarding the safe handling of the chemicals specified in this method.

6.0 Apparatus, Equipment, and Supplies

- 6.1 Segmented Flow Analysis (SFA) System (OI Analytical Flow Solution® IV) consisting of the following:
 - 6.1.1 Model 502 Multichannel Peristaltic Pump
 - 6.1.2 Random Access (RA) Autosampler
 - 6.1.3 Expanded Range (ER) Photometric Detector with 5-mm path length flowcell and 420-nm optical filter
 - 6.1.4 Data Acquisition System (PC or Notebook PC) with WinFLOW™ software
 - 6.1.5 Distillation Module (Part #A515000)
 - 6.1.6 Volatile Acidity in Wine Cartridge (Part #A003133)
- 6.2 Sampling equipment—Sample bottle, amber glass, with polytetrafluoroethylene (PTFE)-lined cap. Clean by washing with detergent and water, rinsing with two aliquots of reagent water, and drying by baking at 110°–150°C for a minimum of one hour.
- 6.3 Standard laboratory equipment including volumetric flasks, pipettes, syringes, etc. should all be cleaned, rinsed, and dried per bottle cleaning procedure in Section 6.2.

7.0 Reagents and Calibrants

- 7.1 Raw Materials
 - 7.1.1 Acetic Acid, glacial, CH_3COOH (FW 60.05)
 - 7.1.2 Brij®-35, 30% w/v (Part #A21-0110-33)
 - 7.1.3 Deionized Water (ASTM Type I or II)
 - 7.1.4 Ethanol, anhydrous, $\text{C}_2\text{H}_5\text{OH}$ (FW 46.07)
 - 7.1.5 Hydrogen Peroxide, 30% w/v, H_2O_2 (FW 34.02)
 - 7.1.6 Potassium Iodate, KIO_3 (FW 214.00)
 - 7.1.7 Potassium Iodide, KI (FW 166.01)
 - 7.1.8 Tartaric Acid, $\text{C}_4\text{H}_6\text{O}_6$ (FW 150.09)